Lithium Hexamethyldisilazide-Mediated Enolizations: Influence of Chelating Ligands and Hydrocarbon Cosolvents on the Rates and Mechanisms

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Supporting Information

Part I. NMR / in situ IR Structural Studies

I	⁶ Li NMR spectra recorded on [6 Li, 15 N]LiHMDS with TMEDA showing enolization of ketone 1 - <i>d</i> ₃ in 2:1 toluene/pentane.
П	⁶ Li NMR spectra recorded on [⁶ Li, ¹⁵ N]LiHMDS with 2.0 equiv DME showing complexation of ketone $1-d_3$ to LiHMDS monomer and dimer in 2:1 toluene/pentane.
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Structures in Supporting Information





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I. ⁶Li NMR spectra of 0.10 M [⁶Li,¹⁵N]LiHMDS with added TMEDA and 1- d_3 in 2:1 toluene/pentane at -100 °C: A) ⁶Li NMR spectrum with 5.0 equiv of added TMEDA; B) ⁶Li NMR spectrum with 5.0 equiv of added TMEDA and 0.2 equiv of added 1- d_3 .



II. ⁶Li NMR spectra of 0.10 M [⁶Li,¹⁵N]LiHMDS with added DME and 1- d_3 in 2:1 toluene/pentane at -100 °C: (A) ⁶Li NMR spectrum with 2.0 equiv of added DME; (B) ⁶Li NMR spectrum with 2.0 equiv of added DME and 0.50 equiv of added 1- d_3 ; (C) ⁶Li {¹⁵N} NMR spectrum of B.



III. ⁶Li NMR spectra of 0.10 M [⁶Li,¹⁵N]LiHMDS with added DME and $1-d_3$ in toluene at -90 °C: (A) ⁶Li NMR spectrum with 10 equiv of added DME; (B) ⁶Li NMR spectrum with 10 equiv of added DME and 0.50 equiv of added $1-d_3$ (showing enolate peaks).



IV. ⁶Li NMR spectra of 0.10 M [⁶Li,¹⁵N]LiHMDS with added *trans*-TMCDA and 1- d_3 in toluene at -90 °C: (A) ⁶Li NMR spectrum with 5.0 equiv added *trans*-TMCDA and 0.2 equiv of added 1- d_3 (before reaction); (B) ⁶Li NMR spectrum of A after warming of tube (showing enolate peaks); (C) ⁶Li {¹⁵N} NMR spectrum of B.



V. ⁶Li NMR spectra of 0.10 M [⁶Li,¹⁵N]LiHMDS with added TMEDA and carbamate **15** in 3.0 M toluene/pentane at -90 °C: (A) ⁶Li{¹⁵N} NMR spectrum with 5.0 equiv of added TMEDA; (B) ⁶Li{¹⁵N} NMR spectrum with 5.0 equiv of added TMEDA and 0.5 equiv of added **15**; (C) ⁶Li{¹⁵N} NMR spectrum with 5.0 equiv of added TMEDA and 1.0 equiv of added **15**; (D) ⁶Li{¹⁵N} NMR spectrum with 5.0 equiv of added TMEDA and 1.0 equiv of added **15**; (D) ⁶Li{¹⁵N} NMR spectrum with 5.0 equiv of added TMEDA and 2.0 equiv of added **15**.



VI. ⁶Li NMR spectra of 0.10 M [⁶Li,¹⁵N]LiHMDS with added *trans*-TMCDA and carbamate **15** in 3.0 M toluene/pentane at -90 °C: (A) ⁶Li{¹⁵N} NMR spectrum with 5.0 equiv of added *trans*-TMCDA; (B) ⁶Li{¹⁵N} NMR spectrum with 5.0 equiv of added *trans*-TMCDA and 0.5 equiv of added **15**; (C) ⁶Li{¹⁵N} NMR spectrum with 5.0 equiv of added *trans*-TMCDA and 1.0 equiv of added **15**; (D) ⁶Li{¹⁵N} NMR spectrum with 5.0 equiv of added *trans*-TMCDA and 2.0 equiv of added **15**.



VII. ⁶Li NMR spectra of 0.10 M [⁶Li,¹⁵N]LiHMDS with added toluene in pentane at -90 °C showing LiHMDS dimer: (A) ⁶Li{¹⁵N} NMR spectrum with 3.0 M added toluene; (B) ⁶Li{¹⁵N} NMR spectrum with 4.0 M added toluene; (C) ⁶Li{¹⁵N} NMR spectrum with 6.0 M added toluene; (D) ⁶Li{¹⁵N} NMR spectrum with 8.0 M added toluene.



VIII. In situ IR spectra recorded on 0.005 M carbamate **15** in 3.0 M toluene/pentane with 0.10 M LiHMDS and 0.50 M TMEDA at (A) 0 °C; (B) -30 °C; (C) -60 °C showing free and complexed carbamate.



IX. In situ IR spectra of ketone $1-d_3$ with 0.10 M LiHMDS and 0.50 M TMEDA in A) neat toluene and B) 2.5 M toluene/pentane at -60 °C.



X. Plot of k_{obsd} vs. [*trans*-TMCDA] for the enolization of **1** (0.005 M) by LiHMDS (0.10 M) in toluene at -55 °C. The curve depicts an unweighted least-squares fit to $k_{obsd} = a[trans$ -TMCDA] + $b (a = 4.6 \pm 0.1 \times 10^{-1}, b = 1.17 \pm 0.02)$.

XI. Table of data for the plot in section I.

[trans-TMCDA] (M)	$k_{\rm obsd} 1 ({\rm s}^{-1})$	$k_{\rm obsd}2~({\rm s}^{-1})$	$k_{\text{obsd}} (\text{avg}) (\text{s}^{-1})$
0.15	$1.37 \pm 0.01\text{E-}3$	$1.07 \pm 0.01 \text{E-}3$	1.2 ±0.2E-3
0.4	1.60 ± 0.01 E-3	1.22 ± 0.01 E-3	$1.4 \pm 0.3 \text{E-}3$
0.6	1.43 ± 0.01 E-3	$1.30 \pm 0.01 \text{E-}3$	$1.40 \pm 0.04 \text{E-}3$
0.9	$1.87 \pm 0.02 \text{E-}3$	$1.46 \pm 0.02\text{E-}3$	$1.6 \pm 0.3 \text{E-}3$
1.2	$1.77 \pm 0.02 \text{E-}3$	$1.62 \pm 0.03 \text{E-}3$	$1.7 \pm 0.1 \text{E-}3$



XII. Plot of k_{obsd} vs. [LiHMDS] for the enolization of **1** (0.005 M) in *trans*-TMCDA (0.40 M) and toluene at -55 °C. The curve depicts an unweighted least-squares fit to $k_{obsd} = a[\text{LiHMDS}]^b$ ($a = 8.0 \pm 0.5$, $b = 0.86 \pm 0.04$).

XIII. Table of data for plot in section XII.

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LiHMDS] (M)	$k_{\rm obsd} 1 \ ({\rm s}^{-1})$	$k_{\text{obsd}}2 \text{ (s}^{-1})$	k_{obsd} (avg) (s ⁻¹)
0.05	$6.17 \pm 0.04 \text{E-4}$	$6.32 \pm 0.03 \text{E-4}$	$6.2 \pm 0.1 \text{E-4}$
0.10	1.16 ± 0.01 E-3	1.22 ± 0.01 E-3	$1.19 \pm 0.04 \text{E-}3$
0.15	$1.42 \pm 0.02 \text{E-}3$	$1.59 \pm 0.03 \text{E-}3$	$1.5 \pm 0.1 \text{E-}3$
0.20	$1.93 \pm 0.04 \text{E-}3$	$1.97 \pm 0.03 \text{E-}3$	$1.95 \pm 0.02E-3$
0.30	2.83 ± 0.08 E-3	$2.97 \pm 0.05 \text{E-}3$	$2.90 \pm 0.09 \text{E-3}$



XIV. Plot of k_{obsd} vs. [TMEDA] for the enolization of 1- d_3 (0.005 M) by LiHMDS (0.10 M) in toluene at -60 °C. The curve depicts an unweighted least-squares fit to $k_{obsd} = a$ [TMEDA] + b ($a = 2 \pm 9 \times 10^{-2}$, $b = 2.4 \pm 0.1$).

XV. Table of data for plot in section **XIV**.

[TMEDA] (M)	$k_{\rm obsd} 1 ({\rm s}^{-1})$	$k_{\rm obsd}2~({\rm s}^{-1})$	$k_{\rm obsd}$ (avg) (s ⁻¹)
0.1	$2.34 \pm 0.05 \text{E-}3$	$2.72 \pm 0.04 \text{E-3}$	2.5 ± 0.1 E-3
0.4	$2.41 \pm 0.05 \text{E-}3$	$2.63 \pm 0.05 \text{E-}3$	$2.5 \pm 0.1 \text{E-}3$
0.9	$2.49 \pm 0.04 \text{E-}3$	2.33 ± 0.05 E-3	$2.4 \pm 0.1 \text{E-}3$
1.4	$2.25 \pm 0.05 \text{E-}3$	$2.45 \pm 0.04 \text{E-}3$	$2.3 \pm 0.1 \text{E-}3$
1.9	$2.77\pm0.05\text{E-}3$	$2.55 \pm 0.06\text{E-3}$	$2.6 \pm 0.1 \text{E-}3$



XVI. Plot of k_{obsd} vs [LiHMDS] for the enolization of $1-d_3$ (0.005 M) in TMEDA (0.40 M) and toluene at -60 °C. The curve depicts an unweighted least-squares fit to $k_{obsd} = a[\text{LiHMDS}]/(1 + b[\text{LiHMDS}])$ ($a = 3.4 \pm 0.6 \times 10^1$, $b = 3.9 \pm 0.1$).

XVII. Table of data for the plot in section XVI.

[LiHMDS] (M)	$k_{\rm obsd} 1 \ ({\rm s}^{-1})$	$k_{\rm obsd}2~({\rm s}^{-1})$	$k_{\text{obsd}}(\text{avg})(\text{s}^{-1})$
0.05	1.59 ± 0.01 E-3	$1.56 \pm 0.01 \text{E-3}$	$1.57 \pm 0.02\text{E-3}$
0.10	$2.41 \pm 0.05 \text{E-}3$	$2.63 \pm 0.05 \text{E-}3$	$2.5 \pm 0.1 \text{E-}3$
0.20	$3.8 \pm 0.1 \text{E-}3$	4.1 ± 0.1 E-3	$3.9 \pm 0.2 \text{E-}3$
0.30	$4.0 \pm 0.1 \text{E-}3$	4.6 ± 0.1 E-3	$4.3 \pm 0.4 \text{E-}3$
0.40	$5.1 \pm 0.3 \text{E-}3$	6.3 ± 0.4 E-3	$5.7 \pm 0.8 \text{E-}3$



XVIII. Plot of k_{obsd} vs [DME] for the enolization of **1** (0.005 M) by LiHMDS (0.10 M) in toluene at -78 °C. The curve depicts an unweighted least-squares fit to $k_{obsd} = a[DME]^b/(1 + c[DME]^b) + d$ ($a = 3.9 \pm 0.3$, $b = -2.6 \pm 0.1$, $c = 3.4 \pm 0.3 \times 10^{-1}$, $d = 9 \pm 1 \times 10^{-1}$.

XIX. Table of data for the plot in section XVIII.

[DME] (M)	$k_{\rm obsd} 1 ({\rm s}^{-1})$	$k_{\rm obsd}2~(\rm s^{-1})$	$k_{\text{obsd}}(\text{avg})(\text{s}^{-1})$
0.0	$8.2 \pm 0.1 \text{E-}3$	$7.6 \pm 0.2 \text{E-}3$	8 ± 1E-3
0.1	1.15 ± 0.01 E-2	$1.23 \pm 0.03 \text{E-}2$	$1.19 \pm 0.05\text{E-}2$
0.4	$1.07 \pm 0.01 \text{E-}2$	$1.03 \pm 0.01 \text{E-}2$	$1.05 \pm 0.02\text{E-}2$
0.6	$6.42 \pm 0.08 \text{E-}3$	$7.66 \pm 0.07 \text{E-}3$	$7.0 \pm 0.8 \text{E-}3$
0.8	$5.2 \pm 0.1 \text{E-}3$	$4.8 \pm 0.1 \text{E-}3$	$5.0 \pm 0.2 \text{E-}3$
1.05	$3.58 \pm 0.05 \text{E-}3$	$3.55 \pm 0.04 \text{E-}3$	$3.56 \pm 0.02\text{E-3}$
1.3	$3.01 \pm 0.05 \text{E-}3$	$2.80 \pm 0.06\text{E-3}$	$2.9 \pm 0.1 \text{E-}3$
1.8	$1.83 \pm 0.02\text{E-}3$	$1.80 \pm 0.02\text{E-}3$	$1.81 \pm 0.02\text{E-3}$
2.3	$1.42 \pm 0.01 \text{E-}3$	$1.51 \pm 0.01\text{E-}3$	$1.46 \pm 0.06\text{E-3}$
2.8	$1.07 \pm 0.01 \text{E-}3$	1.100±0.009E-3	$1.08 \pm 0.02\text{E-3}$
3.8	$9.6 \pm 0.1 \text{E-4}$	$1.04 \pm 0.01 \text{E-}3$	$1.00 \pm 0.05 \text{E-3}$
4.8	$8.75 \pm 0.08 \text{E-4}$	$8.35 \pm 0.07 \text{E-4}$	$8.5 \pm 0.2 \text{E-4}$
5.8	$7.69 \pm 0.08 \text{E-4}$	$7.85 \pm 0.06\text{E-4}$	$7.7 \pm 0.1 \text{E-4}$
6.8	$9.27 \pm 0.06\text{E-4}$	$8.79 \pm 0.09 \text{E-4}$	$9.0 \pm 0.3 \text{E-4}$
7.8	$9.4 \pm 0.1 \text{E-4}$	$9.1 \pm 0.1 \text{E-4}$	$9.2 \pm 0.2 \text{E-4}$



XX. Plot of k_{obsd} vs. [LiHMDS] for the enolization of **1** (0.005 M) in DME (1.0 equiv) and toluene at -78 °C. The curve depicts an unweighted least-squares fit to $k_{obsd} = a$ [LiHMDS] + b ($a = 1.0 \pm 0.6 \times 10^{1}$, $b = 7.3 \pm 0.7$).

XXI. Table of data for the plot in section **XX**.

[LiHMDS] (M)	$k_{\rm obsd} 1 ({\rm s}^{-1})$	$k_{\rm obsd}2~({\rm s}^{-1})$	$k_{\text{obsd}}(\text{avg})(\text{s}^{-1})$
0.05	$7.2 \pm 0.2 \text{E-}3$	$8.80 \pm 0.07 \text{E-}3$	8 ± 1E-3
0.10	$8.2 \pm 1E-3$	$7.6 \pm 0.2 \text{E-}3$	$7.9 \pm 0.4 \text{E-}3$
0.15	9.1 ± 0.3 E-3	$8.9 \pm 0.1 \text{E-}3$	$9.0 \pm 0.1 \text{E-}3$



XXII. Plot of k_{obsd} vs. [LiHMDS] for the enolization of **1** (0.005 M) in DME (1.30 M) and toluene at -78 °C. The curve depicts an unweighted least-squares fit to $k_{obsd} = a[\text{LiHMDS}]/(1 + b[\text{LiHMDS}])$ ($a = 3.1 \pm 0.1 \times 10^{1}$, $b = 1.3 \pm 0.2$).

XXIII. Table of data for the plot in section XXII.

[LiHMDS] (M)	$k_{\rm obsd} 1 \ ({\rm s}^{-1})$	$k_{\rm obsd}2~({\rm s}^{-1})$	$k_{\text{obsd}}(\text{avg})(\text{s}^{-1})$
0.05	1.32 ± 0.01 E-3	$1.55 \pm 0.01\text{E-3}$	$1.4 \pm 0.1 \text{E-}3$
0.10	$3.01 \pm 0.05 \text{E-}3$	$2.80 \pm 0.06\text{E-3}$	$2.9 \pm 0.1 \text{E-}3$
0.20	$4.8 \pm 0.2 \text{E-}3$	$4.8 \pm 0.2 \text{E-}3$	$4.8 \pm 0.0 \text{E-}3$
0.30	$6.24 \pm 0.06\text{E-3}$	$7.0 \pm 0.1 \text{E-}3$	$6.6 \pm 0.5 \text{E-}3$
0.35	$7.2 \pm 0.3 \text{E-}3$	$7.8 \pm 0.2 \text{E-}3$	$7.5 \pm 0.4 \text{E-}3$



XXIV. Plot of k_{obsd} vs. [LiHMDS] for the enolization of **1** (0.005 M) in DME (6.80 M) and toluene at -78 °C. The curve depicts an unweighted least-squares fit to $k_{obsd} = a[\text{LiHMDS}]^b$ ($a = 1.90 \pm 0.07 \times 10^1$, $b = 1.35 \pm 0.2$).

XXV. Table of data for the plot in section **XXIV**.

[LiHMDS] (M)	$k_{\text{obsd}} 1 \text{ (s}^{-1})$	$k_{\rm obsd}2~({\rm s}^{-1})$	$k_{\text{obsd}}(\text{avg})(\text{s}^{-1})$
0.05	$3.61 \pm 0.02\text{E-4}$	3.61 ± 0.03 E-4	3.61E-4
0.10	$9.27 \pm 0.06\text{E-4}$	$8.79 \pm 0.09 \text{E-}4$	$9.0 \pm 0.3 \text{E-4}$
0.15	$1.38 \pm 0.01 \text{E-}3$	$1.42 \pm 0.02E-3$	$1.40 \pm 0.02\text{E-}3$
0.20	$2.18 \pm 0.04 \text{E-}3$	$2.19 \pm 0.03E-3$	$2.185 \pm 0.007\text{E-3}$
0.25	$2.92 \pm 0.09 \text{E-}3$	$2.74 \pm 0.04 \text{E-}3$	$2.8 \pm 0.1 \text{E-3}$
0.30	$3.61 \pm 0.09 \text{E-}3$	$3.89 \pm 0.08\text{E-}3$	$3.7 \pm 0.1 \text{E-}3$
0.35	$4.4 \pm 0.1 \text{E-}3$	$4.8 \pm 0.1 \text{E-}3$	$4.6 \pm 0.2 \text{E-}3$



XXVI. Plot of k_{obsd} vs. [toluene] for the enolization of 1- d_3 (0.005 M) by LiHMDS (0.10 M) in TMEDA (0.50 M) and pentane at -60 °C. The curve depicts an unweighted least-squares fit to $k_{obsd} = a$ [toluene]^b ($a = 1.5 \pm 0.3 \times 10^2$, $b = -1.6 \pm 0.1$).

XXVII. Table of data for the plot in section XXVI.

[toluene] (M)	$k_{\rm obsd} 1 \ ({\rm s}^{-1})$	$k_{\rm obsd}2~({\rm s}^{-1})$	$k_{\text{obsd}}(\text{avg})(\text{s}^{-1})$
3.00	$2.45 \pm 0.03 \text{E-}2$	$2.25 \pm 0.06\text{E-}2$	2.3 ± 0.1 E-2
4.26	$1.57 \pm 0.03 \text{E-}2$	$1.41 \pm 0.02\text{E-}2$	$1.5 \pm 0.1 \text{E-}2$
5.68	$8.2 \pm 0.1 \text{E-}3$	$8.8 \pm 0.1 \text{E-3}$	$8.5 \pm 0.4 \text{E-}3$
7.00	$4.5 \pm 0.1 \text{E-}3$	$5.2 \pm 0.2 \text{E-}3$	$4.8 \pm 0.4 \text{E-}3$
8.52	$2.8 \pm 0.1 \text{E-3}$	$2.97\pm0.06\text{E-}3$	$2.8 \pm 0.1 \text{E-}3$



XXVIII. Plot of k_{obsd} vs. [mesitylene] for the enolization of $1-d_3$ (0.005 M) by LiHMDS (0.10 M) in TMEDA (0.50 M) and pentane at -65 °C. The curve depicts an unweighted least-squares fit to $k_{obsd} = a$ [mesitylene]^b ($a = 1.3 \pm 0.3 \times 10^2$, $b = -1.8 \pm 0.1$).

XIX. Table of data for the plot in section XXVIII.

[mesitylene] (M)	$k_{\rm obsd} 1 \ ({\rm s}^{-1})$	$k_{\text{obsd}}2 \text{ (s}^{-1})$	$k_{\text{obsd}}(\text{avg})(\text{s}^{-1})$
3.50	$1.38 \pm 0.03\text{E-}2$	$1.18 \pm 0.03\text{E-}2$	1.2 ± 0.1 E-2
4.50	$8.5 \pm 0.2 \text{E-}3$	$8.3 \pm 0.2 \text{E-}3$	$8.4 \pm 0.1 \text{E-}3$
5.68	$4.97 \pm 0.08 \text{E-}3$	$4.9 \pm 0.1 \text{E-}3$	$4.93 \pm 0.04 \text{E-}3$



XXX. Plot of k_{obsd} vs [toluene] for the enolization of **1** (0.005 M) by LiHMDS (0.10 M) in TMCDA (0.50 M) and pentane at -55 °C. The curve depicts an unweighted least-squares fit to $k_{obsd} = a$ [toluene]^b ($a = 7.2 \pm 0.4 \times 10^{1}$, $b = -1.81 \pm 0.04$).

XXXI. Table of data for the plot in section XXX.

[toluene] (M)	$k_{\rm obsd} 1 ({\rm s}^{-1})$	$k_{\text{obsd}}2 \text{ (s}^{-1})$	$k_{\text{obsd}}(\text{avg})(\text{s}^{-1})$
3.00	$9.9 \pm 0.6 \text{E-}3$	-	$9.9 \pm 0.6 \text{E-}3$
4.17	$5.5 \pm 0.2 \text{E-}3$	-	$5.5 \pm 0.2 \text{E-}3$
6.00	$2.92 \pm 0.05 \text{E-}3$	-	$2.92 \pm 0.05 \text{E-}3$
8.34	$1.22 \pm 0.01 \text{E-}3$	$1.60 \pm 0.01 \text{E-}3$	$1.4 \pm 0.3 \text{E-}3$