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I. 6Li, 15N, and 13C NMR spectra of 0.10 <u>M</u> [6Li,15N]LiHMDS with added Et₂O at -100 °C: (A) 6Li NMR spectrum with 0.7 equiv. of added Et₂O in pentane; (B) 15N{1H} NMR spectrum with 0.7 equiv. of added Et₂O in pentane; (C) 6Li NMR spectrum with 40 equiv. of added Et₂O in pentane; (D) 15N{1H} NMR spectrum with 40 equiv. of added Et₂O in pentane; (E) 13C NMR spectrum with 0.5 equiv. of added Et₂O in toluene-d₈; (F) 13C NMR spectrum with 2.0 equiv. of added Et₂O in toluene-d₈.



II. ⁶Li, ¹⁵N, and ¹³C NMR spectra of 0.10 <u>M</u> [⁶Li,¹⁵N]LiHMDS with added 2,2-Me₂THF at -100 oC: (A) ⁶Li NMR spectrum with 0.7 equiv. of added 2,2-Me₂THF in pentane; (B) ¹⁵N{¹H} NMR spectrum with 0.7 equiv. of added 2,2-Me₂THF in pentane; (C) ⁶Li NMR spectrum with 40 equiv. of added 2,2-Me₂THF in pentane; (D) ¹⁵N{¹H} NMR spectrum with 40 equiv. of added 2,2-Me₂THF in pentane; (D) ¹⁵N{¹H} NMR spectrum with 40 equiv. of added 2,2-Me₂THF in pentane; (E) ¹³C NMR spectrum with 0.5 equiv. of added 2,2-Me₂THF in toluene-d₈; (F) ¹³C NMR spectrum with 2.0 equiv. of added 2,2-Me₂THF in toluene-d₈.





ppm

ppm



IV. ⁶Li, ¹⁵N, and ¹³C NMR spectra of 0.10 <u>M</u> [⁶Li,¹⁵N]LiHMDS with added *i*-PrOMe at -100 °C: (A) ⁶Li NMR spectrum with 0.7 equiv. of added *i*-PrOMe in pentane; (B) ¹⁵N{¹H} NMR spectrum with 0.7 equiv. of added *i*-PrOMe in pentane; (C) ⁶Li NMR spectrum with 40 equiv. of added *i*-PrOMe in pentane; (D) ¹⁵N{¹H} NMR spectrum with 40 equiv. of added *i*-PrOMe in pentane; (E) ¹³C NMR spectrum with 0.5 equiv. of added *i*-PrOMe in toluene-d₈; (F) ¹³C NMR spectrum with 2.0 equiv. of added *i*-PrOMe in toluene-d₈.



V. ⁶Li, ¹⁵N, and ¹³C NMR spectra of 0.10 <u>M</u> [⁶Li,¹⁵N]LiHMDS with added *t*-BuOMe: (A) ⁶Li NMR spectrum with 0.7 equiv. of added *t*-BuOMe in pentane at -120 °C; (B) ⁶Li NMR spectrum with 40 equiv. of added *t*-BuOMe in pentane at -100 °C; (C) ¹⁵N{¹H} NMR spectrum with 40 equiv. of added *t*-BuOMe in pentane at -100 °C; (D) ¹³C NMR spectrum with 0.5 equiv. of added *t*-BuOMe in toluene-d₈ at -110 °C (E) ¹³C NMR spectrum with 2.0 equiv. of added *t*-BuOMe in toluene-d₈ at -110 °C. (An ¹⁵N NMR spectrum of 0.7 equiv. of added *t*-BuOMe was not recorded due to solubility problems.)



VI. 6Li, 15N, and 13C NMR spectra of 0.10 <u>M</u> [6Li,15N]LiHMDS with added *n*-BuOMe at -100 °C: (A) 6Li NMR spectrum with 0.7 equiv. of added *n*-BuOMe in pentane; (B) 15N{1H} NMR spectrum with 0.7 equiv. of added *n*-BuOMe in pentane; (C) 6Li NMR spectrum with neat *n*-BuOMe; (D) 15N{1H} NMR spectrum with neat *n*-BuOMe in pentane; (E) 13C NMR spectrum with 0.5 equiv. of added *n*-BuOMe in toluene-d₈; (F) 13C NMR spectrum with 2.0 equiv. of added *n*-BuOMe in toluene-d₈.

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VII. ⁶Li, ¹⁵N, and ¹³C NMR spectra of 0.10 <u>M</u> [⁶Li,¹⁵N]LiHMDS with added THP at -100 °C: (A) ⁶Li NMR spectrum with 0.7 equiv. of added THP in pentane; (B) ¹⁵N{¹H} NMR spectrum with 0.7 equiv. of added THP in pentane; (C) ⁶Li NMR spectrum with 40 equiv. of added THP in pentane; (D) ¹⁵N{¹H} NMR spectrum with 40 equiv. of added THP in pentane; (E) ¹³C NMR spectrum with 0.5 equiv. of added THP in toluene-d₈; (F) ¹³C NMR spectrum with 2.0 equiv. of added THP in toluene-d₈.



VIII. ⁶Li, ¹⁵N, and ¹³C NMR spectra of 0.10 <u>M</u> LiHMDS with added oxetane at -100 °C: (A) ⁶Li NMR spectrum of [⁶Li,¹⁵N]LiHMDS with 0.7 equiv. of added oxetane in pentane; (B) ⁶Li NMR spectrum of [⁶Li]LiHMDS with 1.1 equiv. of added oxetane and 2.0 equiv. of added THF in toluene-d₈; (C) ⁶Li NMR spectrum of [⁶Li,¹⁵N]LiHMDS with 13 equiv. of added oxetane in pentane; (D) ¹⁵N{¹H} NMR spectrum of [⁶Li,¹⁵N]LiHMDS with 13 equiv. of added oxetane in pentane; (E) ¹³C NMR spectrum of [⁶Li]LiHMDS with 2.0 equiv. of added oxetane in toluene-d₈; (F) ¹³C NMR spectrum of [⁶Li]LiHMDS with 2.0 equiv. of added oxetane in toluene-d₈.





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X. Partial ¹³C NMR spectra of 0.10 <u>M</u> [⁶Li]LiHMDS in toluene-d₈ with 5.0 equiv. of Et₂O per Li recorded at -100 °C: (A) 0.0 equiv. THF; (B) 0.25 equiv. THF; (C) 0.5 equiv. THF; (D) 0.75 equiv. THF; (E) 1.0 equiv. THF.



XI. ¹³C{1H} NMR spectra of 0.10M [⁶Li]LiHMDS with Et₂O and added ethereal solvent in toluene-d₈ at -100 °C: (A) 1.1 equiv. of added *i*-PrOMe and 3.0 equiv of added Et₂O; (B) 4.0 equiv. of added *t*-BuOMe and 1.1 equiv of added Et₂O;



XI.(cont) (C) 1.1 equiv. of added 2,2-Me₂THF and 5.0 equiv. of added Et_2O ; (D) 1.1 equiv of added 2-MeTHF and 5.0 equiv. of added Et_2O ;





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XII. ¹³C{¹H} NMR spectra of 0.10M [⁶Li]LiHMDS with THF and added ethereal solvent in toluene-d₈ at -100 °C: (A) 3.0 equiv. of added 2,2-Me₂THF and 1.1 equiv of added THF; (B) 4.0 equiv. of added *n*-BuOMe and 1.1 equiv of added THF;



XII.(cont.) (C) 1.1 equiv. of added THP and 1.1 equiv. of added THF; (D) 2.0 equiv of added 2-MeTHF and 1.1 equiv. of added THF.



XIII. 6Li-15N HMQC spectra of 0.10 <u>M</u> [6 Li, 15 N]LiHMDS in pentane at -100 °C: (A) 0.5 equiv. of added 2-MeTHF and 1.0 equiv. of added Et₂O; (B) 0.5 equiv. of added THF and 1.0 equiv. of added 2-MeTHF; (C) 0.5 equiv. of added *n*-BuOMe and 1.0 equiv. of added Et₂O; (D) 0.5 equiv. of added THF and 1.0 equiv. of added *n*-BuOMe. The upper and left-hand traces are the corresponding ^{15}N {¹H} and ^{6}Li NMR spectra.



XIV. ⁶Li-¹⁵N HMQC spectra of 0.10 <u>M</u> [⁶Li,¹⁵N]LiHMDS in pentane at -100 °C: (A) 0.5 equiv. of added THF and 1.0 equiv. of added *i*-PrOMe; (B) 0.5 equiv. of added *i*-PrOMe and 1.0 equiv. of added Et₂O; (C) 0.5 equiv. of added THF and 1.0 equiv. of added 2,2-Me₂THF; (D) 0.5 equiv. of added 2,2-Me₂THF and 1.0 equiv. of added Et₂O;



XIV.(cont.) (E) 0.5 equiv. of added THF and 0.7 equiv. of added THP; (F) 0.5 equiv. of added THP and 1.0 equiv. of added Et₂O; (G) 0.5 equiv. of added THF and 1.0 equiv. of added Et₂O (H) 0.5 equiv. of added Et₂O and 1.0 equiv of added *t*-BuOMe at -120 °C. The upper and left-hand traces are the corresponding $^{15}N{^{1H,6}Li}$ and ^{6}Li NMR spectra, respectively.



XV. ⁶Li-detected ¹⁵N zero-quantum NMR spectra of 0.1 <u>M</u> [⁶Li,¹⁵N]LiHMDS in pentane at -100 °C: (A) 5 equiv. of Et₂O; (B) 5 equiv. of *t*-BuOMe; (C) 5 equiv. 2-MeTHF; (D) 5 equiv. 2,2-Me₂THF;



XV.(cont.) (E) 40 equiv. 2,2,5,5-Me₄THF; (F) 5 equiv. THP; (G) 5 equiv. *i*-PrOMe; (H) 5 equiv. *n*-BuOMe.



XV.(cont.) (I) 5 equiv. Me₂(Et)COMe; (J) 40 equiv. *i*-Pr₂O; (K) 5 equiv. oxetane; (L) 5 equiv. THF



XV.(cont.) (M) neat pentane.



XVI. Variable temperature ¹³C spectra of 0.1 <u>M</u> [⁶Li]LiHMDS with added Et₂O: (A) 5.0 equiv. of added Et₂O; (B) 2.0 equiv. of added Et₂O (C) 1.2 equiv. of added Et₂O.



XVII. Variable temperature ¹³C spectra of 0.1 <u>M</u> [⁶Li]LiHMDS with added THF: (A) 5.0 equiv. of added THF; (B) 2.0 equiv. of added THF (C) 1.2 equiv. of added THF.



XVIII. Plot of $[AS_n]/[A_2S_2]^{1/2}$ vs. [2-MeTHF] for 0.1 <u>M</u> LiHMDS in pentane. The data are fit by non-linear least squares methods to the function in equation 10 of the manuscript: (A) at -20 °C K_{eq} = 7.8 x 10⁻³, n = 2.9; (B) at -80 °C, K_{eq} = 2.9 x 10⁻³, n = 3.1.

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XIX. (A) Plot of $[AS_n]/[A_2S_2]^{1/2}$ vs. $[2,2-Me_2THF]$ for 0.1 <u>M</u> LiHMDS in pentane at -80 °C. The data are fit by non-linear least squares methods to the function in equation 10 of the manusript. $K_{eq} = 6.0 \times 10^{-2}$, n = 3.1. (B) Plot of $[AS_n]/[A_2S_2]^{1/2}$ vs. [oxetane] for 0.1 <u>M</u> LiHMDS in pentane at -20 °C. The data are fit by non-linear least squares methods to the function in equation 10 of the manuscript. $K_{eq} = 2.3 \times 10^{-2}$, n = 3.7.



XX. Predicted concentrations of disolvated dimer 1I, trisolvated monomer 12I and tetrasolvated monomer 13I at -80 °C in oxetane. The functions are calculated using adjustable parameters $K_{eq(1)} = 7.1 \times 10^{-3}$ and $K_{eq(2)} = 6.8 \times 10^{-1}$ derived from nonlinear least squares fit to equation 12.

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Solvent	∆H∘ _f (11)	∆H° _f (10)	∆H∘ _f (S)
THF	-398.4	-257.2	-59.3
2-MeTHF	-398.9	-258.9	-62.5
2,2-Me ₂ THF	-390.9	-252.7	-61.8
Et ₂ O	-390.6	-250.9	-62.0
<i>n-</i> BuOMe	-406.0	-262.3	-66.0
<i>i</i> -PrOMe	-384.4	-246.8	-58.2
t-BuOme	-371.7	-236.8	-54.6
THP	-399.0	-258.1	-60.1
i-Pr ₂ O	-379.7	-240.3	-64.5
H₂O	-415.6	-270.0	∘60.9
Me ₂ O	-382.0	-240.8	-51.2

XXI. Tabulated thermodynamic data from MNDO semiempirical calculations on disolvated dimer (11) and disolvated monomer (10).^a

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^aEnthalpies are reported in kcal/mol.

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XXII. Plot of observed LiHMDS aggregation free energies ($\Delta\Delta G_{agg}^{\circ}$, equation 2 and 3, Table 3) vs LiHMDS aggregation enthalpies ($\Delta\Delta H_{agg(calc)}^{\circ}$, equation 6) calculated from MNDO.



XXIII. Plot of LiHMDS monomer solvation enthalpies ($\Delta\Delta H^{o}_{solv(calc)}$ (monomer), equation 7) calculated from MNDO vs. calculated LiHMDS dimer solvation enthalpies ($\Delta\Delta H^{o}_{solv(calc)}$ (dimer), equation 8).

Given the equilibria

$$\begin{array}{ccc} & \mathsf{K}_1 & \mathsf{K}_2 \\ 1/2 \,\mathsf{A}_2 \,\mathsf{S}_2 + 3 \mathrm{S} & \rightleftharpoons \mathrm{A} \,\mathsf{S}_3 + \mathrm{S} & \rightleftharpoons \mathrm{A} \,\mathsf{S}_4 \end{array}$$

such that

$$K_{1} = \frac{[AS_{3}]}{[A_{2}S_{2}]^{1/2}[S]^{2}}$$
(1)

and

$$K_{2} = \frac{[AS_{4}]}{[AS_{3}][S]}$$
(2)

we can derive the equations describing the equilibrium constants as a function of solvent and organolithium concentrations. We define the total monomer concentration, A_T , such that

 $[A_T] = [AS_3] + [AS_4]$

Substituting into equation 2 and rearranging affords

$$[AS_3] = \frac{[A_T]}{(K_2)[S] + 1}$$

Squaring equation 1, substituting for [AS₃], and rearranging affords

$$\frac{[A_T]}{[A_2S_2]} = K_1^2[S]^4((K_2)[S]+1)^2$$
(3)

Since the total LiHMDS concentration equals 0.10 M, then

$$[A_2S_2] = \frac{0.10 - [A_T]}{2}$$

Substituting into equation 3 for [A2S2] and rearranging affords

$$[A_T]^2 + \frac{K_1^2[S]^4((K_2)[S] + 1)^2[A_T]}{2} - \frac{(0.1) K_1^2[S]^4((K_2)[S] + 1)^2}{2} = 0$$

Solving for [AT] using the quadratic equation affords

$$[A_{T}] = \frac{K_{1}^{2}[S]^{4}((K_{2})[S] + 1)^{2}}{4} + \frac{[S]^{2}((K_{2})[S] + 1)}{2} \sqrt{\frac{K_{1}^{4}[S]^{4}((K_{2})[S] + 1)^{2}}{2} + (0.2)K_{1}^{2}}$$



XXV. $^{6}Li^{-15}N$ HMQC spectrum of 0.10 <u>M</u> [^{6}Li , ^{15}N]LiHMDS with 0.3 equiv. THF per Li. The left and upper traces are the corresponding one-dimensional ^{6}Li and ^{15}N { ^{6}Li , ^{1}H }NMR spectra, respectively.

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